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# **Mass patterning of polysiloxane layers using spin coating and photolithography techniques**

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## **Abstract**

Polysiloxane (PSX) containing 2,2-dimethoxy-2-phenylacetophenone (DMPA) photoinitiator has been used as a photosensitive polymer. Thus, thin PSX films have been deposited by spin coating and patterned thanks to standard ultraviolet (UV) photolithography. The influences of the different technological parameters (PSX dilution, spin speed, UV exposure time) have been studied in so as to understand the polysiloxane deposition and cross-linking phenomena. Finally, the whole process has been optimised. Results evidence the realisation of high quality PSX patterns for the development of mass-fabricated ion sensitive layers in the field of chemical microsensors.

**Keywords:** polysiloxane copolymer, spin coating, photolithography

## Introduction

Within the framework of microtechnologies, polymers seem very promising materials. Indeed, their multitude gives access to a wide range of multidisciplinary properties in order to bring solutions to many technological problems. The field of chemical sensors has not been an exception to the rule. Thus, photosensitive polymers based on poly-vinyl alcohol (PVA), poly-vinyl chloride (PVC), polymethacrylate (PolyHEMA) or polysiloxane (PSX), have been used for the preparation of chemical or biochemical sensitive membranes by manual dip coating and exposure to ultraviolet (UV) radiation [1-5]. However, irrespective of the polymer chosen, such processes have not been fully reproducible or/and reliable. In order to assess the chemical sensors industrial development, mass-fabrication processes deviated from silicon microelectronics have to be carried out. Solutions involve the development of spin coating and photolithography techniques even if the spinning process is known to be responsible for the loss of large amounts of chemical products. Nevertheless, they have been successfully developed for the deposition and mass patterning of thin poly-vinyl alcohol PVA layers [6] and their use should be extended to other photosensitive polymers.

This paper deals with the development of spin coating and ultraviolet (UV) photolithography techniques applied to standard photosensitive polysiloxane (PSX) copolymer. The different technological parameters (PSX dilution, spin speed, UV exposure time) have been studied in so as to understand the photosensitive PSX copolymer deposition and UV cross-linking phenomena. Thus, the whole process has been optimised for the mass patterning of polysiloxane-based ion sensitive layers in the field of chemical microsensors.

## Experiments

The polysiloxane (PSX) copolymer RMS-033 containing methacrylate groups was purchased from ABCR. The 3-(methacryloxy)propyl trimethoxysilane (MPTS), the 2,2-dimethoxy-2-phenylacetophenone (DMPA) photoinitiator, the tetrahydrofuran (THF) solvent as well as the others standard chemical products used were obtained from Sigma-Aldrich.

According to the Warsaw University of Technology [5], the photosensitive polymer was formed from 99% of PSX and 1% of DMPA, and then dissolved into THF to reach different dilutions  $x_0$ . The volumic mass was of the different PSX solutions obtained was finally estimated by accurate weighing while their hydrodynamics properties, and more precisely their static viscosity, were characterized using a Haake VT550 viscometer. All the data related to the different PSX solutions were finally regrouped in table 1.

The polysiloxane thin films were patterned according to the following process. After an MTPS/toluene silanisation step at 80°C to promote adherence on the silicon substrate, the PSX-based solution was deposited by spin coating using a SUSS GYRSET RC-8 spin coater. Then, after the THF solvent evaporation under nitrogen atmosphere, ultraviolet (UV) exposure was performed under a non-oxidising ambient using a Karl Suss MA-6 contact alignment system. Finally, after development into xylene, the PSX pattern thicknesses and profiles were studied using a TENCOR scan profiler. Thus, different spin speeds and UV exposure times were tested in order to understand fully the polysiloxane cross-linking phenomena and to optimize the whole technological process.

## Results and discussion

In order to understand the polysiloxane cross-linking phenomena during the photolithography process, the influence of the UV exposure has been studied for different spin

coating steps and for the different PSX solutions. Figures 1 and 2 represent respectively the variations of the polysiloxane thickness  $h$  (after the development step into xylene) as a function of the UV exposure time  $t_{UV}$  for different spin speeds  $\omega$  and for the different PSX solutions, i.e. for the different dilutions into THF  $x_0$ . As expected, experimental results have demonstrated that the PSX thickness decreases with both increases of spin speed and PSX dilution whatever the UV exposure time. Furthermore, in all cases, the PSX thickness  $h$  follows exponential temporal variations according to the following equation:

$$h(t) = h_{\text{dep}} \left[ 1 - \exp\left(-\frac{t}{\tau}\right) \right] \quad (1)$$

The time constant  $\tau$  has been studied as a function of the different technological parameters concerned. Thus, it appears that it does not depend on the spin speed  $\omega$  but decreases linearly with the PSX dilution into THF  $x_0$  (figure 3). This result shows that the PSX cross-linking due to the photolithography process depend mainly on the polymer physicochemical properties. Furthermore, such exponential variations demonstrate that these cross-linking phenomena under UV exposure goes from the substrate interface towards the film surface. Therefore, a minimal UV exposure time  $t_{\text{min}}$  is necessary to reach the nominal deposited PSX thickness  $h_{\text{dep}}$ . Whatever the PSX solution and the spin coating step, this  $t_{\text{min}}$  value has never been found lower than 240s.

Then, the nominal deposited PSX thickness  $h_{\text{dep}}$  has been studied as a function of the spin speed  $\omega$  for different polysiloxane dilution into THF  $x_0$  (figure 4). It has been ranging between 10 and 40 micrometers and, as already said, decreases with both increases of the spin speed and the PSX dilution. To have a better understanding of these variations, the hydrodynamics of the spin-coating process must be applied to the PSX/THF solution. In the case of a Newtonian polymer

solution applied to the horizontal surface of a circular substrate, Meyerhofer has demonstrated that the deposited thickness  $h_{\text{dep}}$  is related to the spin speed  $\omega$  according to the following relation [7]:

$$h_{\text{dep}} = x_0 \left( \frac{3v_0 e}{2(1-x_0)\omega^2} \right)^{\frac{1}{3}} \quad (2)$$

with:  $v_0 = \eta_0/\rho_0$

where  $x_0$  is the PSX dilution into THF,  $v_0$ ,  $\eta_0$  and  $\rho_0$  are the kinematic viscosity, the static viscosity and the volumic mass of the PSX polymer solution respectively (see table 1), and  $e$  is the THF surfacic evaporation per unit time .

If agreement between Meyerhofer theory and experience was found in the case of PVA/water solutions [6], it has not been truly obtained in the case of the PSX/THF solutions. Indeed, according to figure 4, the PSX thickness varies roughly with  $\omega^{-5/6}$  rather than  $\omega^{-2/3}$  (equation 2). This discrepancy can be explained in different ways. Firstly, the PSX/THF solution could not be a Newtonian polymer solution. Secondly, the theoretic assumptions done by Meyerhofer in order to take account the evaporation phenomena could also not be fully adapted to the studied PSX/THF solution [7]. Thirdly, the polysiloxane copolymer RMS-033 supplied by ABCR is initially diluted in an “unknown” solvent (for confidential reasons...) which may interfere during the spin-coating process and be finally responsible for the misfit between experience and theory. Finally, this misfit could be simply related to the dependence of the THF surfacic evaporation per unit time  $e$  with the spin speed  $\omega$ , and more precisely with  $\omega^{-1/2}$ .

Nevertheless, the whole technological process has been optimised and validated through the mass-fabrication of micrometric polysiloxane patterns (dimensions:  $1000 \times 1000 \times 15 \mu\text{m}^3$  and  $800 \times 100 \times 15 \mu\text{m}^3$ ) with a good homogeneity and a good reproducibility of fabrication (figures 5

and 6). These developments have put into evidence drawbacks related to the photolithography step. For too high UV exposure time  $t_{UV}$ , UV diffraction phenomena are responsible for the obtaining of badly defined PSX patterns. Solution has been found by decreasing the UV exposure time while keeping the PSX deposited thickness roughly equal to its maximal value  $h_{dep}$  (see equation 1). Thus, after optimisation, the UV diffraction phenomena have finally been found to be responsible for a 15% dimensional increase at the pattern base as well as for a 5% thickness heterogeneity near the pattern edge (figures 5 and 6). If these results have to be still improved, they are reasonable for the development of ion sensitive layers in the field of chemical sensors.

## **Conclusion**

The mass patterning of photosensitive polysiloxane (PSX) layers has been investigated. A technological process based on spin coating and ultraviolet photolithography techniques has been proposed. PSX deposition, cross-linking and diffraction phenomena have been studied and understood even if some discrepancies with the spin-coating deposition hydrodynamic theory have been evidenced. Finally, optimisation of the whole technological process has lead to the fabrication of high quality, micrometric, polysiloxane patterns while achieving a good homogeneity and a good reproducibility of fabrication. These technological works will be used for the development of ion sensitive layers for chemical sensors such as ion sensitive field effect transistors (ISFETs).

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## **Tables and figures captions**

Table 1: Characteristics of the different PSX solutions

Figure 1: PSX thickness  $h$  as a function of UV exposure time  $t_{UV}$  for different PSX solutions (spin speed:  $\omega = 2500$  rounds per minute)

Figure 2: PSX thickness  $h$  as a function of UV exposure time  $t_{UV}$  for different spin speeds  $\omega$  (solution S2, PSX dilution into THF:  $x_0 = 0.83$ )

Figure 3: time constant  $\tau$  as a function of PSX dilution into THF  $x_0$  for different spin speeds  $\omega$

Figure 4: nominal PSX thickness  $h_{dep}$  as a function of spin speed  $\omega$  for different PSX solutions

Figure 5: optimized PSX pattern (dimensions:  $1000 \times 1000 \times 15 \mu\text{m}^3$ ) picture and profile

Figure 6: optimized PSX pattern (dimensions:  $800 \times 100 \times 15 \mu\text{m}^3$ ) picture and profile

PSX solution	S1	S2	S3	S4
PSX/DMPA (99/1) volume (mL)	15.15	15.15	15.15	15.15
THF volume (mL)	1	3	5	7
dilution $x_0$	0.94	0.83	0.75	0.68
volumic mass $\rho$ (kg.m <sup>-3</sup> )	983	972	963	956
static viscosity $\eta$ (kg.m <sup>-1</sup> .s <sup>-1</sup> )	1.7	0.9	0.47	0.3
kinematic viscosity $\nu$ (x 10 <sup>-3</sup> m <sup>2</sup> .s <sup>-1</sup> )	1.73	0.93	0.49	0.31

Table 1: Characteristics of the different PSX solutions

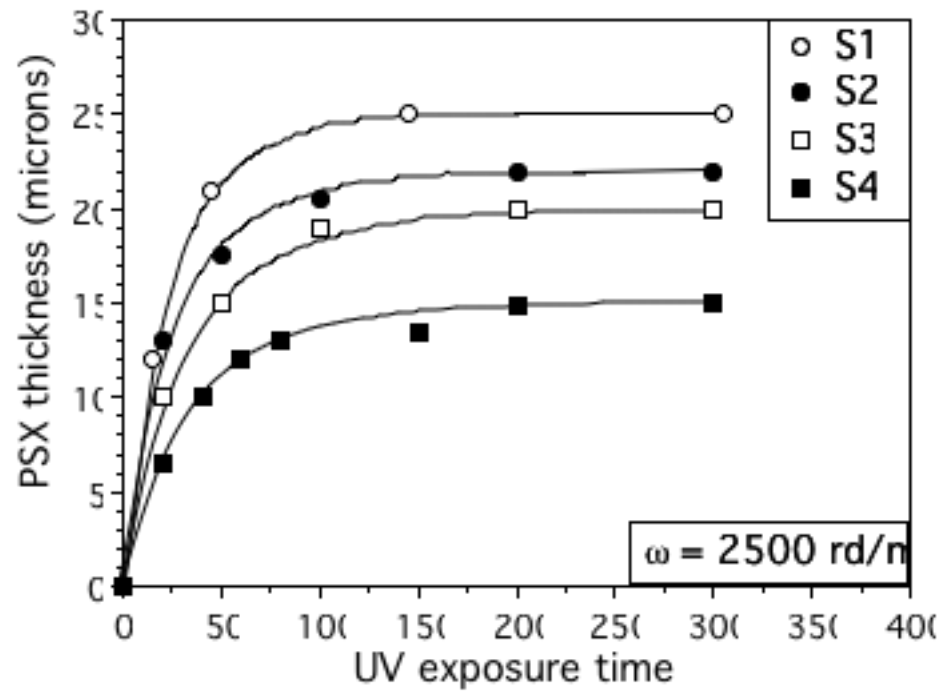


Figure 1: PSX thickness  $h$  as a function of UV exposure time  $t_{UV}$  for different PSX solutions (spin speed:  $\omega = 2500$  rounds per minute)

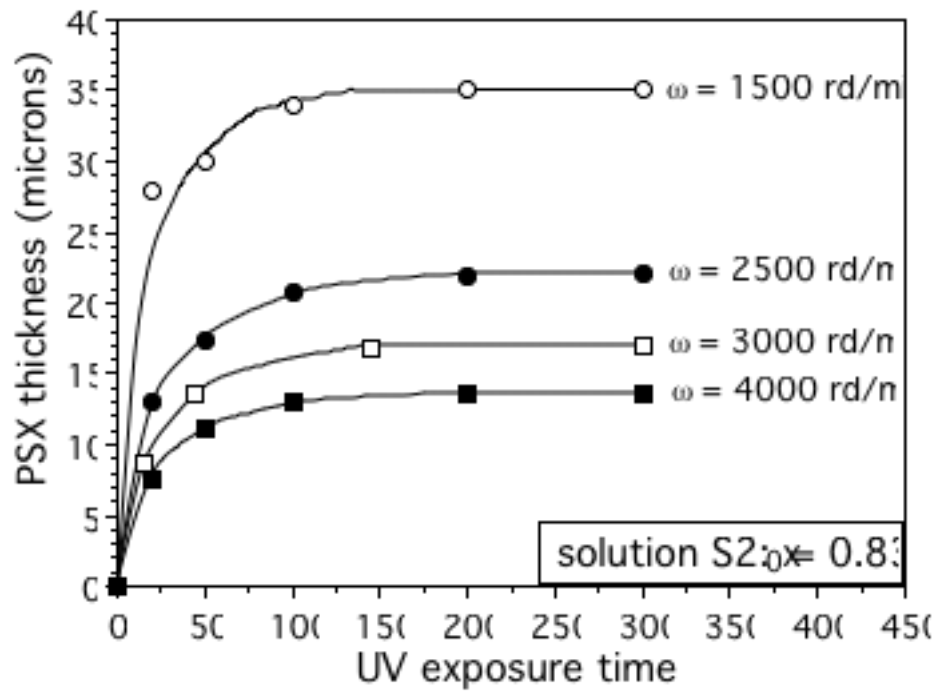


Figure 2: PSX thickness  $h$  as a function of UV exposure time  $t_{UV}$  for different spin speeds  $\omega$  (solution S2, PSX dilution into THF:  $x_0 = 0.83$ )

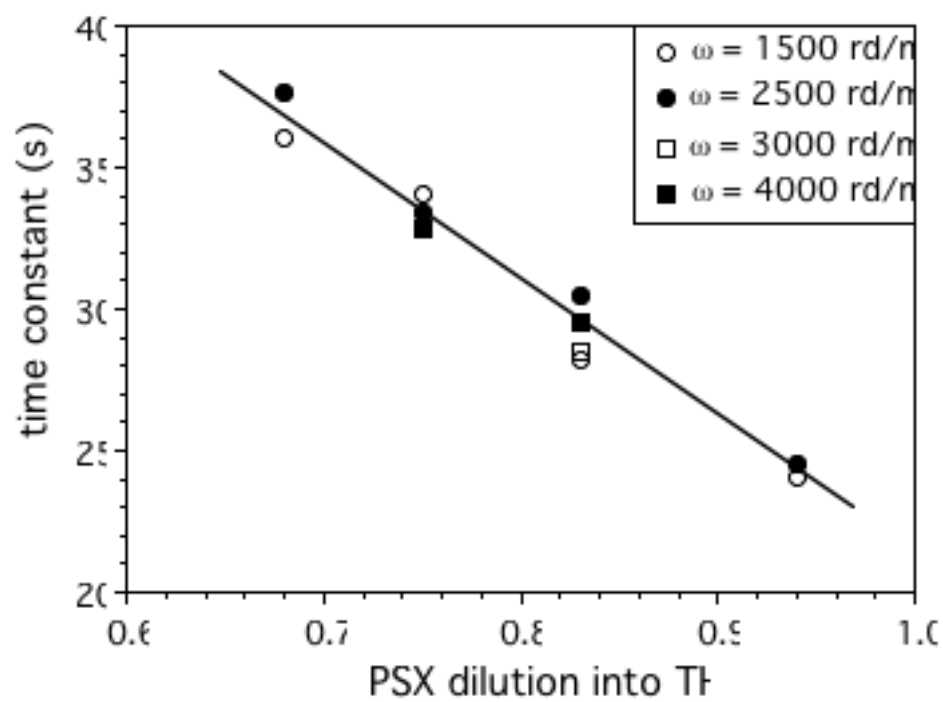


Figure 3: time constant  $\tau$  as a function of PSX dilution into THF  $x_0$  for different spin speeds  $\omega$

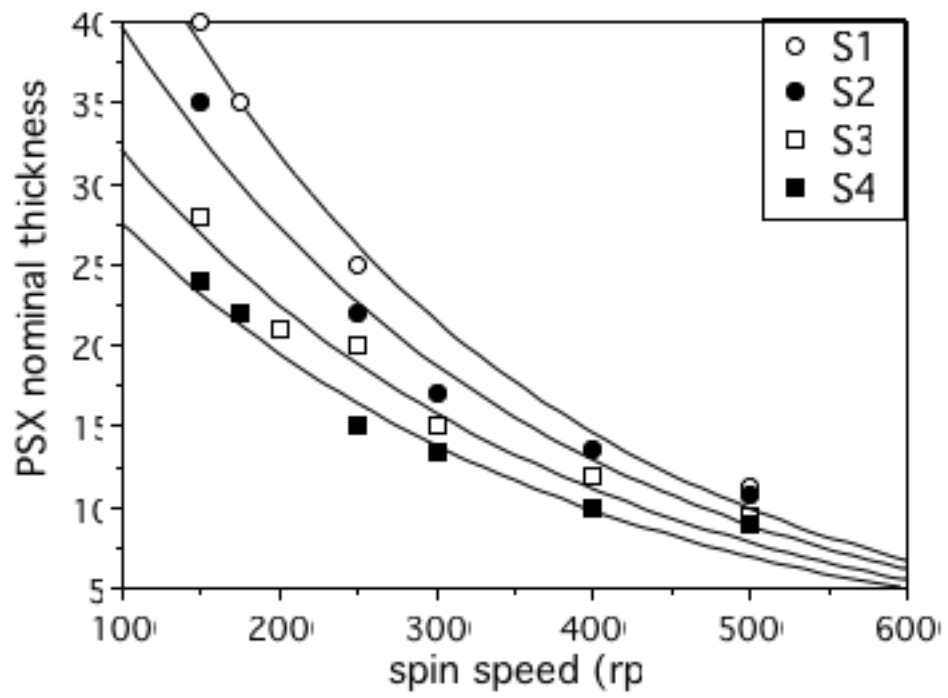


Figure 4: deposited thickness  $h_{\text{dep}}$  as a function of spin speed  $\omega$  for different PSX solutions

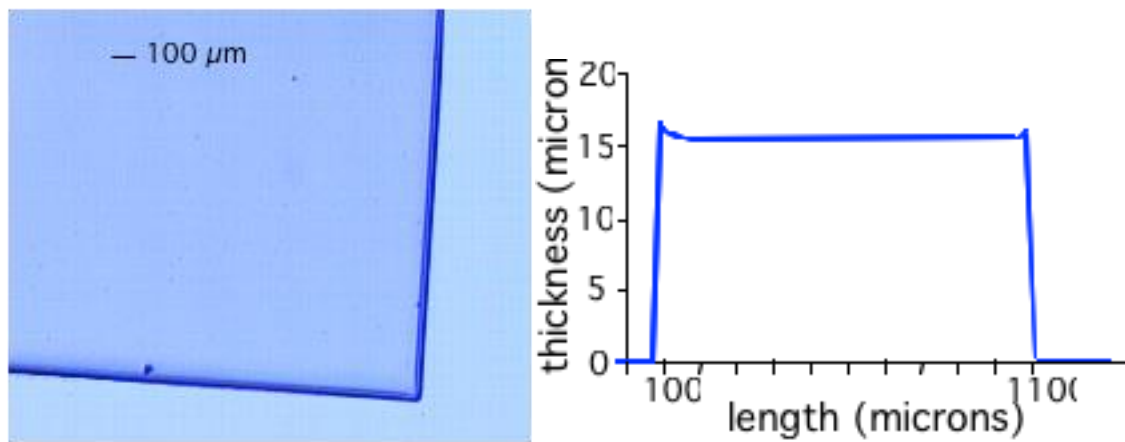


Figure 5: optimized PSX pattern (dimensions:  $1000 \times 1000 \times 15 \mu\text{m}^3$ ) picture and profile

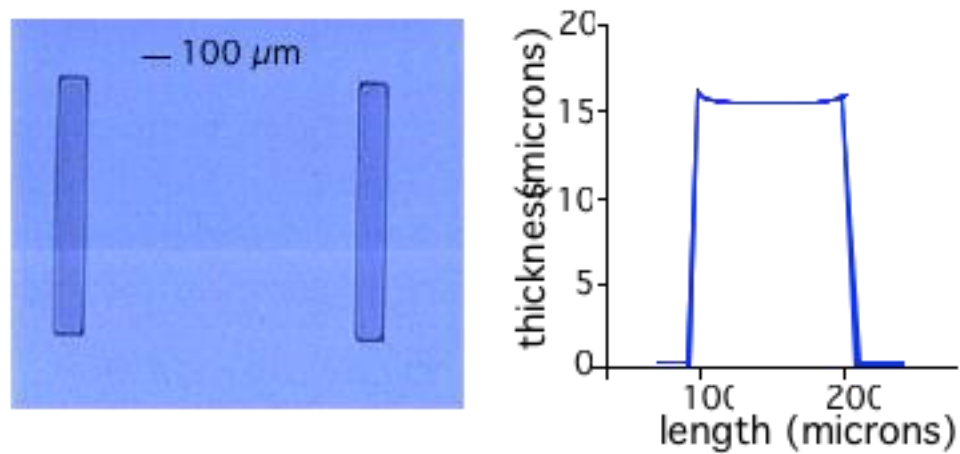


Figure 6: optimized PSX pattern (dimensions:  $800 \times 100 \times 15\ \mu\text{m}^3$ ) picture and profile