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Proceedings

Polyethylene Outgassing Study for MEMS Nuclear Radiation Sensor

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Abstract: In this paper, the out-gassing from PolyEthylene (PE) film is studied under nuclear irradiation for the design and fabrication of passive MEMS dosimeters. The fabrication of sensor needs high temperature process that leads to specific constraints on the PE film. Radiation chemical yield of hydrogen production (G_{H2}) from PE under gamma irradiation is verified by mass spectroscopy after temperature annealing in vacuum up to 400°C. Prototypes are fabricated to validate the fabrication of the sensor and then irradiated with high energy gamma radiation (with dose of 20 kGy). Measurements of membrane deflections after irradiation validate the G_{H2} factor, showing low pollution level of PE during sensor fabrication.

Keywords: polyethylene; nuclear radiation sensor; MEMS dosimeters; hydrogen outgassing

1. Introduction

Wireless passive (battery-less) and chip-less (without electronic circuits) sensors present interesting solution for harsh environment applications. In the case of nuclear radiation measurement, such solutions allow real time dose monitoring by remotely interrogating nuclear radiation sensors from radar reader [1].

The operating principle of the sensor is the following: when PolyEthylene (PE) placed inside a hermetic sealed cavity is subjected to nuclear irradiation, an outgassing (mainly of H_2 molecules) occurs and generates a pressure on a highly resistive silicon membrane. This lead to the membrane deflection, which modify the resonant frequency of a planar microwave resonator placed just below the membrane. From the measurement of the frequency shift, the nuclear radiation dose is derived.

In our previous design (see Figure 1), a double Si/Glass stack was used for the sensor fabrication and leads to a complex technological process and non-sealed cavity for the resonator. The PE was obtained from peeling small ball and the Si/Glass2 bonding was performed under N_2 using a non-collective process Moreover, the sensor sensitivity varied significantly from one sensor to another and was about 20 times higher than the simulation results [2]. The possible reason could be the pollution of PE during sensor fabrication leading to the undesirable out-gassing after irradiation. Consequently, in order to control these additional out-gassing of PE, we report here a simple, more reliable and collective technological fabrication process for passive MEMS dosimeters with a new design (see Figure 2). The process uses a simple Si/Glass stack where the calibrated 200 μ m thick High Density PolyEthylene (HDPE) film and the resonator are sealed inside the same cavity under vacuum. Moreover, the millimeter-wave excitation of the resonator is performed by using a small radiating aperture [1].

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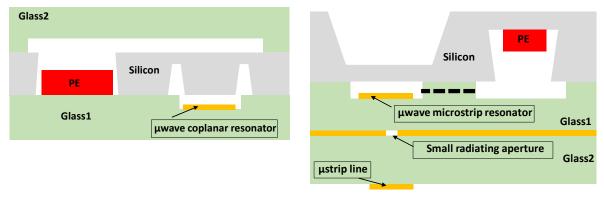


Figure 1. Previous dosimeter design

Figure 2. New dosimeter design

2. Materials and Methods

Prototypes were designed to measure radiation chemical yield of hydrogen production (G_{H2}) from sensing devices fabricated with technological process similar to that used for the final sensors (see Figure 3). The HDPE cavity (5mm diameter and 290 μ m depth) is etched inside a low resistivity 400 μ m thick SOI wafer. The H_2 out-gassing will provide the deflection of the small silicon membrane (550 μ m diameter and 5 μ m thick) that will be monitored with optical profiler.

For G_{H2} = 4.10⁻⁷ mol/J, the pressure P_g generated by the PE is given by equation 1 [2], where V_{PE} and V_g denote respectively the PE volume and the volume of gas expansion:

$$P_g[Bar] = 9.5 \ 10^{-3} \ \frac{V_{PE} \ [mm^3]}{V_g \ [mm^3]} \cong 10^{-2} \ \frac{V_{PE} \ [mm^3]}{V_g \ [mm^3]} \ /kGy$$
 (1)

With dimensions given in the Figure 3, $P_g \cong 0.045$ bar/10kGy which leads a membrane deflection sensitivity of $0.23\mu m/10kGy$.

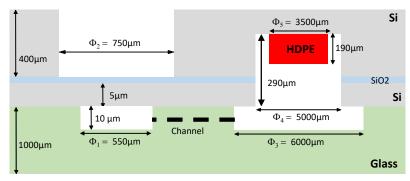


Figure 3. Cross section of the prototype for the measurement of G_{H2}

3. Results and discussions

3.1. PE properties

One of the major issue during the technological fabrication of sensors is the use of high temperature silicon/glass anodic bonding that may degrade the H_2 out-gassing capabilities of HDPE. The thermo-gravimetric analysis performed under Helium (He) atmosphere shows that the degradation of HDPE occurs only above 400°C (see Figure 4). To provide a security margin, the silicon/glass temperature bonding is then performed at 350°C.

HDPE samples are annealed in vacuum at 350°C and then placed in a sealed glass tube under He atmosphere. After irradiation to gamma rays (137 Cs 661.7 keV, dose rate 0.3kGy/h, dose 24kGy), the gas composition analysis is finally performed using a quantitative gas mass spectrometer. It shows that H₂ production is not affected by high temperature process ($G_{H2} \cong 4 \ 10^{-7} \ mol \ / \ J$) [3].

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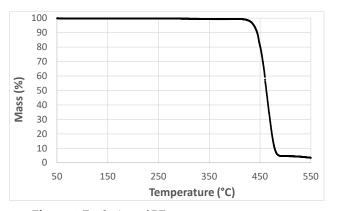


Figure 4. Evolution of PE mass versus temperature

3.2. Prototypes fabrication and characterization before irradiation

For silicon wafer fabrication, the two main steps were: (1) the 5μ m thick membrane release and (2) the HDPE cavity fabrication with deep reactive ionic etching using 40μ m thick photoresist mask. For the glass wafer, the 10μ m depth cavity were obtained from etching in hydrofluorydric bath using Chromium/Gold mask.

After 2 minutes annealing at 200°C under vacuum, 200µm HDPE film was cut using a stamping pliers and the 3.5mm diameter samples were placed inside the silicon cavity. Silicon and glass wafer were aligned in AML bonder, heated at 350°C under vacuum during 30min and then bonded at 600V. Prototypes were fabricated using this technological process (see Figure 5). Before silicon/glass bonding the membrane deflection is small (< 0.2 µm) and after bonding the deflection is found to

bonding, the membrane deflection is small ($\approx 0.2 \mu m$) and after bonding, the deflection is found to range from $5.1 \mu m$ to $5.4 \mu m$ for atmospheric pressure applied on the top of the membrane (see Figure 6). This experimental result is close to the expected value with vacuum inside the cavity (simulated sensitivity was of $5 \mu m/bar$). After two weeks, the membrane deflection remains unchanged.

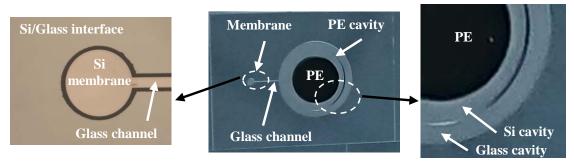


Figure 5. Bottom views of fabricated prototypes

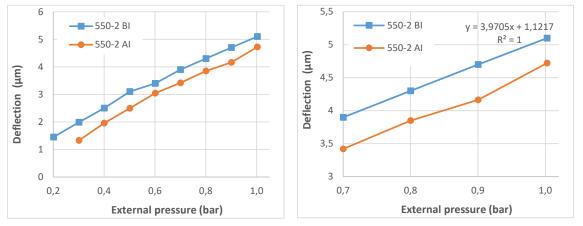


Figure 6. Membrane deflection versus external pressure applied on the top of the membrane (BI: Before Irradiation, AI: After Irradiation)

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As displayed in Figure 6, the membrane sensitivity to pressure variation was also calibrated under pressure between 0.2 bar and 1 bar. For pressure between 0.8 bar and 1 bar, the sensitivity S is around 4 μ m/bar.

3.3. Prototypes characterization after irradiation

The prototypes are irradiated to gamma rays (137 Cs 661.7 keV, dose rate 0.3kGy/h, dose 20kGy). The membrane sensitivity to pressure was also calibrated 3 weeks after irradiation and we checked that no significant change occurred after irradiation (see Figure 6). For atmospheric pressure, the membrane deflection is reduced by 0.4 μ m. The pressure P_g generated inside the HDPE cavity after this irradiation can be then derived from equation (2):

$$P_{q} = \Delta P_{o} - (\Delta D/S) \tag{2}$$

where ΔP_o denote the atmospheric pressure variation between two measurements of membrane deflection, ΔD is the variation of the membrane deflection related to irradiation dose and S designates the membrane sensitivity around 1 bar derived from Figure 6. The pressure P_g is then found to be of 0.05 bar/10kGy and is close to the expected value (simulation results predict a pressure of 0.045 bar/10kGy).

After two weeks, the membrane deflection remains unchanged and consequently, we concluded that a good H₂ tightness was achieved.

4. Conclusions

Prototypes based on silicon membrane were designed and fabricated to validate the fabrication process of a passive MEMS dosimeter. After irradiation to gamma rays (20kGy), membrane deflection analysis shows that the radiation chemical yield of hydrogen production (G_{H2}) from HDPE is close to simulation results. So few undesirable out-gassing from polyethylene occurs during prototype irradiation, validating the fabrication process with low level of HDPE pollution.

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