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High-resolution on-chip supercapacitors with ultra-high scan rate ability

Ty Mai Dinh,^{ab} Kevin Armstrong,^c Daniel Guay^c and David Pech^{*ab}

A simple approach for the fabrication of high-resolution on-chip supercapacitors with increased performances is demonstrated. The resulting micro-devices display high specific cell capacitance (3 mF cm^{-2} using hydrous ruthenium oxide electrode) and impressive scan rate abilities, up to $10\,000 \text{ V s}^{-1}$ for multi-walled carbon nanotubes based micro-supercapacitors, which is five orders of magnitude higher than conventional supercapacitors.

The tremendous push towards systems on chip (SoC) approach together with the decrease of their dimensions has enabled the development of new applications such as wireless sensors network or wearable devices that can be integrated into everyday-use objects. As microelectronic devices get smaller and smaller, it has become critical to power them with micro-power sources of appropriate size, fulfilling specific requirements for self-powered applications such as long lifetime and high-rate charge/discharge capabilities. Electrochemical capacitors (also called supercapacitors) show very interesting characteristics when it comes to these applications, because of their extended lifetimes and high power densities. Supercapacitors store electrical energy using the electrochemical double layer formed at the electrode/electrolyte interface (electrochemical double layer capacitance) or the very fast and reversible electrochemical reaction of some materials deposited at the surface of an electrode (pseudo-capacitance).^{1,2} The realization of on-chip micro-sized supercapacitors could satisfy a variety of micro-power demands and embedded microsystems.³

We have recently shown that the power ability of micro-supercapacitors can be dramatically improved by reducing the physical separation between the anode and the cathode.^{4,5} Indeed, an optimized configuration with very high resolution patterns should considerably increase the performances of micro-supercapacitors (capacitance, energy and power normalized to its footprint area on the chip) due to reduced ohmic losses, better impregnation by the electrolyte, higher accessibilities and compactness. However, most of “micro-supercapacitors” reported in literature have resolution above $100 \mu\text{m}$, the term “micro” referring mainly to the micro-fabrication techniques used for their fabrication.⁶⁻¹¹ Regardless of the

deposition technique, the common challenge is to deposit porous materials on micro-sized substrates without shorting the positive and negative electrodes.¹²

Here, we demonstrate a new and simple approach for large-scale fabrication of micro-supercapacitors with resolution (*i.e.* spacing between interdigitated fingers) down to the limit of standard photolithography, *i.e.* $1 \mu\text{m}$ (see Fig. S1 in ESI†). The process has been successfully applied for the realization of electric double-layer micro-capacitors (μ -EDLCs) based on multi-walled carbon nanotubes (MWCNTs), and pseudo-capacitive electrochemical micro-capacitors (μ -SCs) based on hydrated ruthenium oxide (hRuO_2). The resulting micro-supercapacitors exhibit excellent electrochemical performances with specific cell capacitance up to 3 mF cm^{-2} for hRuO_2 and life cycle exceeding 30 000 cycles for MWCNTs. All micro-devices display also, impressive scan rate abilities, up to $10\,000 \text{ V s}^{-1}$ for MWCNTs, which is five orders of magnitude higher than conventional supercapacitors. The proposed strategy is promising for batch fabrication and integration of operational on-chip micro-supercapacitors into existing micro systems. With electron beam or nano-imprint lithography, the process could also be used for the realization of nano-supercapacitors with sub-micrometer pattern resolutions.

The basic idea for the realization of μ -SCs with high resolution is illustrated in Fig. 1a,b. Current collectors (Ti/Au in this case) were first patterned onto an oxidized silicon wafer and protected by an insulating SU8 masking layer using conventional photolithography techniques. We focused, then, on a lift-off process to realize the micro-devices. This is a widely used method for transferring patterns from resist layers to the material to be structured. In our case, we used this technique to pattern, with high resolution, the current collectors as well as the active material.¹³ The material deposition was then facilitated as it occurs at wafer level, which is the key to batch fabrication of functional materials on chips. Most importantly, however, this method avoids the necessity to deposit the electrode material onto micro-sized patterned current collectors – which often leads to deposition on unwanted areas. We validated the process down to $1 \mu\text{m}$ using two different deposition techniques: the

electrophoretic deposition of MWCNTs and the electrolytic deposition of hRuO₂ (see Fig. S2 in ESI†). For practical reasons, the micro-devices were realized with a 10 μm interspace. For both materials, an adherent layer was obtained with a well-defined pattern, without any short circuit between the electrodes (Fig. 1c,d). The thicknesses, which can be easily modulated by controlling the time used for the electrodeposition of the active layer, were 260 and 100 nm for MWCNTs and hRuO₂, respectively. As illustrated in Fig. 2, this process has overcome resolution limits for the integration of active materials into micro-fabrication technologies, with one of the smallest resolution ever to be applied for “micro-supercapacitor” applications.^{14–17} We have investigated the influence of such resolutions on the performances of $\mu\text{-SCs}$.

Cyclic voltammograms (CVs) were recorded at scan rates ranging from 0.05 to 10 000 V s^{-1} in a de-aerated 0.5 M H₂SO₄ electrolyte to evaluate the power capability of the micro-devices. The micro-supercapacitors of both materials display the rectangular shape voltammograms expected for double layer and pseudo-capacitive materials. The symmetrical capacitive behavior was maintained up to an ultrafast scan rate of 10 V s^{-1} for the hRuO₂-based $\mu\text{-SC}$ (see Fig. S3 in ESI†), and to an impressive value of 1 000 V s^{-1} for the MWCNTs-based $\mu\text{-SC}$ (Fig. 3). The MWCNTs micro-sized device was even able to cycle at an ultrafast scan rate of 10 000 V s^{-1} , which is more than five orders of magnitude higher than conventional supercapacitors, and, to the best of our knowledge, the highest value ever reported for micro-supercapacitors.^{4,18–21} The specific capacitances were 0.2 and 3.2 mF cm^{-2} at 10 V s^{-1} for the MWCNTs and the hRuO₂-based $\mu\text{-SCs}$, respectively. These values are comparable to the specific capacitances reported in the literature at much lower scan rates for $\mu\text{-EDLCs}$ (0.4 - 1.5 mF cm^{-2})^{22–24} and pseudo-capacitive $\mu\text{-SCs}$ (0.4 - 12.6 mF cm^{-2}).^{5,12,25} The high scan rate, representative of a high instantaneous power, is thus concomitant with a high specific capacitance. This ability to cycle at such scan rates is due to the very low value of the time constant (τ) of the interdigitated micro-supercapacitors ($\tau = RC$, where ‘ C ’ is the cell capacitance and ‘ R ’ is the equivalent series resistance, ESR), thanks to their very high resolution pattern.

To highlight the beneficial effect of the resolution on the power performance, we have simulated the influence of the interspace (i) between the electrodes of planar interdigitated $\mu\text{-SCs}$ on their time constant (τ), according to the following equation:

$$\tau = (R_{cc} + K \times \rho_e)C$$

with R_{cc} the resistance of the current collectors, ρ_e the resistivity of the electrolyte and K the analytical expression of the cell constant developed by W. Olthuis *et al.*^{5,26}

As shown in Fig. 4a, a micro-scale reduction of the interspace between adjacent fingers (of 100 μm width and 1 000 μm length) significantly reduces the contribution of the ESR and consequently, the time constant (τ) is diminished. The corresponding simulations of the CV curves at 1 000 V s^{-1} for various time constants, according to the equations developed by B.E. Conway,²⁷ are shown Fig. 4b. We can clearly see that the CVs of the high resolution micro-supercapacitors can keep a rectangular shape at scan rates (s) exceeding 1 000 V s^{-1} . It is noteworthy that the scan rate ability only depends on the RC time constant.

Fig. 5a,b shows the Nyquist and Bode plots of the MWCNTs-based $\mu\text{-SC}$ from 100 kHz to 10 mHz. No leakage current is observed, with the near-vertical straight line (in the low-frequency region) typical of a capacitive material. A negligible equivalent distributed resistance EDR (due to the thin thickness of the active material) and a very low ESR of 0.16 $\Omega \text{ cm}^2$ were estimated from the high frequency part of the spectrum. As the specific power will predominantly depend on the highest resistive contribution between the ESR and the EDR, the influence of the resolution of the configuration of the micro-device is prominent, in this case, with a maximum specific power of 1.3 W cm^{-2} . Taking into account the capacitance, the MWCNTs-based $\mu\text{-SC}$ is characterized by an extremely low time constant (τ) of 32 μs (512 μs for the hRuO₂-based $\mu\text{-SC}$, see Fig. S4 in ESI†), which is in complete agreement with the theoretically determined values.

The electrochemical stability of the deposited layers was finally investigated (Fig. 6 and Fig. S5 in ESI†). The capacitance slightly increases after 30 000 cycles at 1 V s^{-1} for the $\mu\text{-EDLC}$, due to a better electrolyte impregnation with time and cycling.

In conclusion, we have realized high-energy micro-supercapacitors based on hydrated ruthenium oxide, and ultrahigh-power micro-supercapacitors based on multi-walled carbon nanotubes using a simple and integrated-circuits-compatible approach for the deposition of the active material, with extremely high resolutions. This process can be applied to other materials or other deposition techniques. The micro-devices offer truly outstanding performances, with discharge rates of up to 10 000 V s^{-1} . The use of conventional UV-photolithography techniques (1 μm resolution) currently limits the size of the pattern, but further work will consider using this process with the stepper projection lithography (350 nm resolution) and electron beam lithography (sub-10 nm resolution) for further reduction of the interspace. In turn, these improvements in electrode design should lead to unprecedented performances of micro-supercapacitors.

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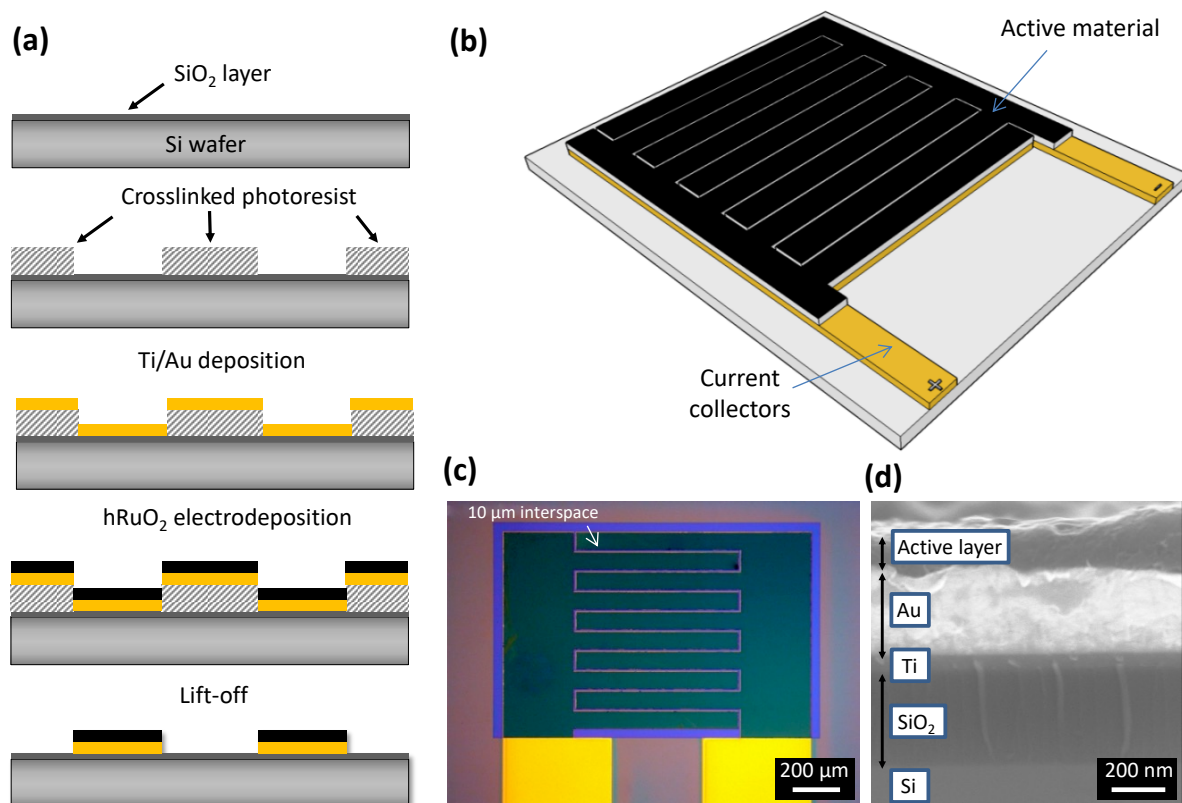


Fig. 1 Micro-supercapacitor fabrication and design. (a) Simplified illustration of the fabrication process using the lift-off process of the active material for high-resolution pattern (not to scale). Schematic (b) and optical image (c) of the micro-device. (d) Scanning electron microscope image of the cross-section of the electrode.

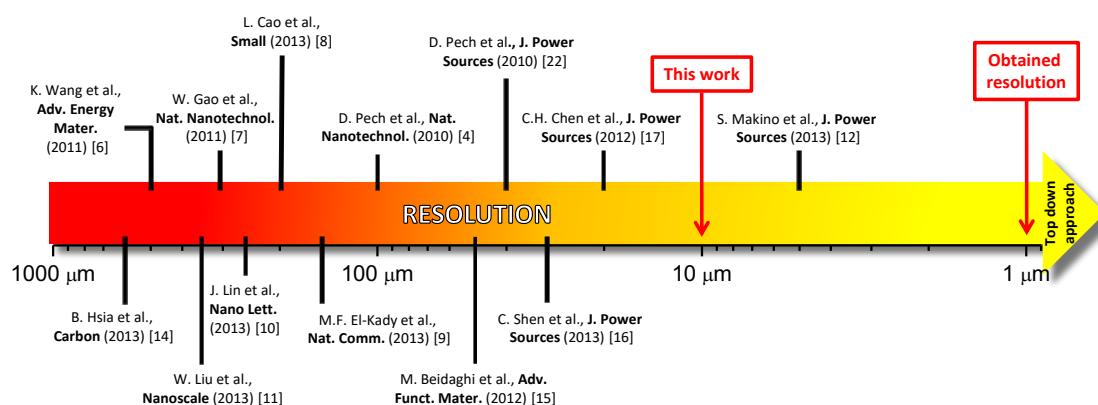


Fig. 2 Comparison of the resolution of state-of-the-art micro-supercapacitors reported in literature.

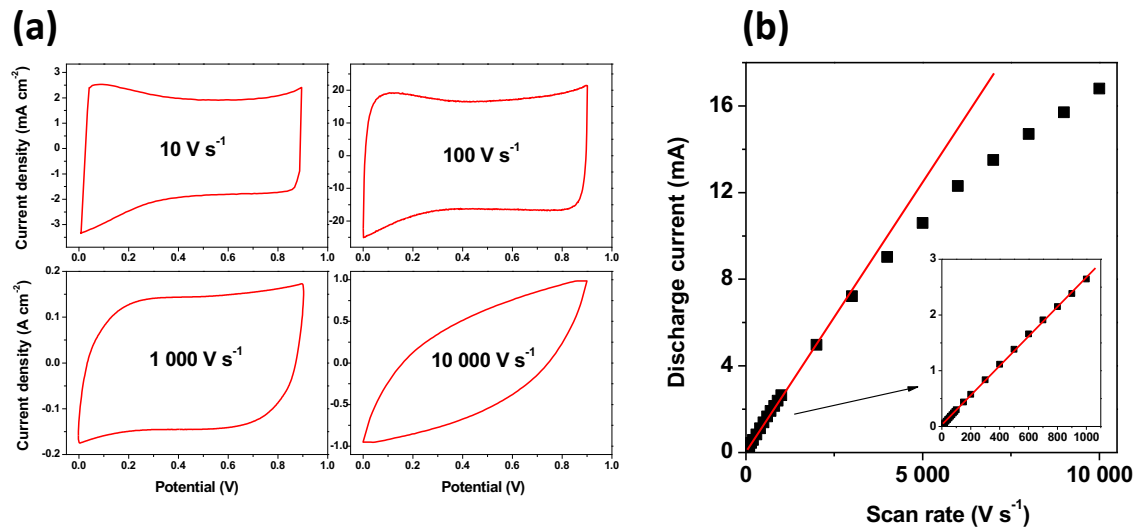


Fig. 3 Cyclic voltammetry characterizations (CV) of MWCNTs-based micro-supercapacitor. (a) CVs obtained at different scan rates in de-aerated $0.5\text{M H}_2\text{SO}_4$ electrolyte. (b) Evolution of the discharge current versus scan rate.

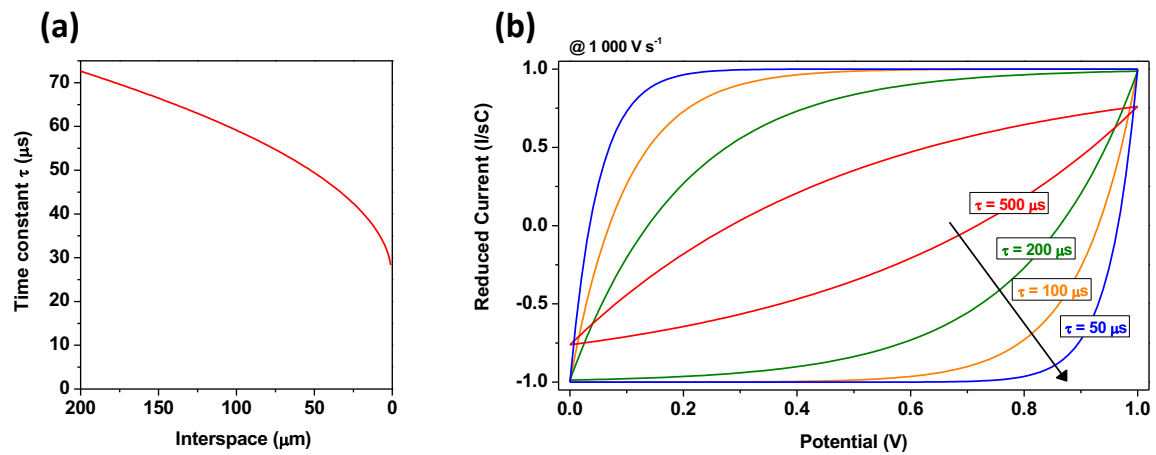


Fig. 4 (a) Theoretical evolution of the time constant τ of MWCNTs-based micro-supercapacitor as a function of the distance between adjacent interdigitated electrodes in $0.5\text{M H}_2\text{SO}_4$. (b) Calculated cyclic voltammograms expressed as reduced response current (I/sC) at 1000 V s^{-1} for various time constant.

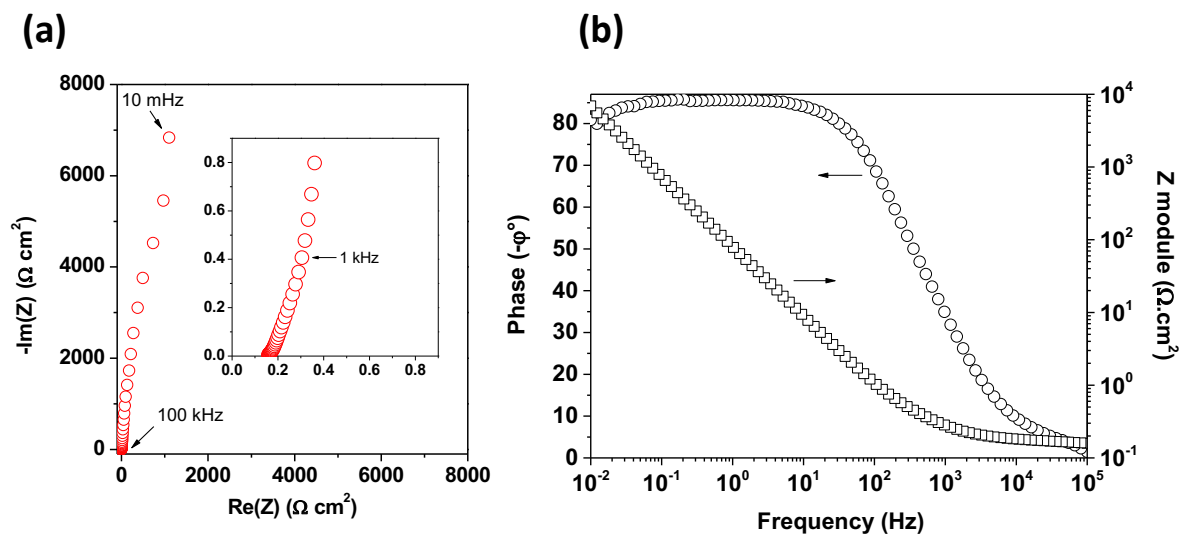


Fig. 5 (a) Nyquist and (b) Bode plots obtained from a MWCNTs-based micro-supercapacitor in 0.5M H_2SO_4 .

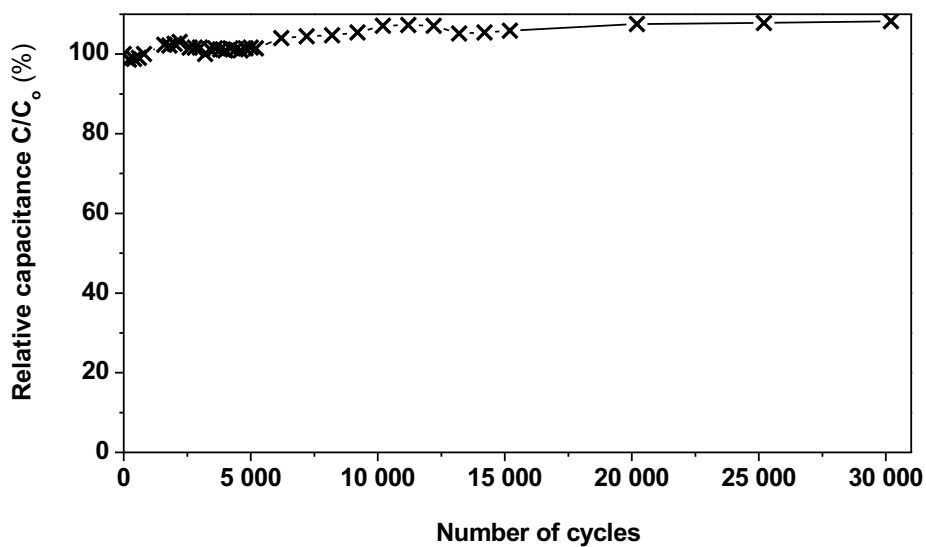


Fig. 6 Evolution of the relative capacitance as a function of the number of electrochemical cycles at 1 V s^{-1} for a MWCNTs-based micro-supercapacitor (C_0 refers to the capacitance obtained for the first cycle).